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(71) Applicant: MANVILLE SALES CORPORATION [US/ US]; Manville Plaza, 5th Floor, P.O. Box 5108, Denver, CO 80217 (US).

(72) Inventors: OLDS, Leonard, Elmo ; 977 South Lake Guich Road, Castle Rock, CO 80104 (US). KIELMEYER, William, Henry; 3374 West Chenango Avenue, Englewood, CO 80110 (US).

(74) Agent: SCHRAMM. William. J.: Brooks & Kushman. 2000 Town Center. Suite 2000. Southfield. MI 48075 (US).

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(54) Title: PROCESS FOR DECOMPOSING AN INORGANIC FIBER

(57) Abstract

Inorganic fibers which have a silicon extraction of greater than 0.02 wt% Si/day in physiological saline solutions. The fiber contains SiO2, MgO, CaO, and at least one of Al2O3, ZrO2, TiO2, B2O3, iron oxides, or mixtures thereof. Also disclosed are norganic fibers which have diameters of less than 3.5 microns and which pass the ASTM E-119 two hour fire test when processed into a fiber blanket having a bulk density in the range of about 1.5 to 3 pcf.

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PROCESS FOR DECOMPOSING AN INORGANIC FIBER

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FIELD OF INVENTION ---

This invention relates to inorganic fiber compositions and more particularly it relates to inorganic fiber compositions which can contain silica, magnesia, calcium oxide, alumina, and other oxides. Some of the inventive fibers have excellent fire ratings, some have especially low durabilities in physiological saline solutions, and some have combinations of these foregoing properties.

BACKGROUND OF THE INVENTION

For many years, inorganic fibers generically referred to in the industry as "mineral wool fibers", made from slag, rock, fly ash, and other by-product raw materials have been manufactured. These fibers have been typically manufactured by melting the slag, rock, etc., containing such oxides as silica, alumina, iron oxide (ferrous and ferric), calcium oxide, and magnesia; allowing the molten material to be blown by gas or steam or to impinge on rotors at high speeds; and causing the resulting blown or spun fibers to be accumulated on a collecting surface. These fibers are then used in bulk or in the form of mates, blankets, and the like as both low and high temperature insulation. U.S. Patent No. 2,576,312 discloses a conventional mineral wool composition and method for making the same.

In the past, the industry has well recognized the standard drawbacks associated with conventional mineral wool fibers. Conventional mineral wool fibers may have high contents of undesired oxides which often

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detract from their refractory properties. The conventional mineral wools are coarse, i.e. they have average fiber diameters of 4 to 5 microns (measured microscopically) and have high shot contents in the range of 30 to 50 weight percent. The coarseness of the fiber reduces the insulating value of the fiber and makes conventional mineral wool unpleasant to handle and unfriendly to the For example, because of their coarse fiber diameters, conventional mineral wool blankets must have bulk densities of from 4 to 8 pcf and even higher in order to pass the ASTM E-119 two hour fire test. On the other hand, fiber glass blankets are often made with bulk densities of 2 pcf or lower. While the fiber glass blankets are friendly because of their low bulk densities and relatively fine fiber diameter, they do not have sufficient fire resistance so as to pass even the one hour ASTM E-119 fire test.

Recently, another potential problem with traditional mineral wool and other types of fiber has been recognized. It is well known that inhalation of certain types of fiber can lead to elevated incidence of respiratory disease, including cancers of the lung and surrounding body tissue. Several occurrences are welldocumented in humans for several types of asbestos Although for other varieties of natural and manmade mineral fiber direct and unequivocal evidence for respiratory disease is lacking, the potential for such occurrence has been inferred from results of tests on laboratory animals. In the absence or insufficiency of direct human epidemiological data, results from fiber inhalation or implantation studies on animals provides the best "baseline information" from which to extrapolate disease potential.

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Chronic toxicological studies on animals have, however, been able to statistically demonstrate the importance of three key factors that relate directly to the potential for respiratory disease and especially carcinoma: (a) dose of fiber received (including time of exposure); (b) dimension of the inhaled fiber; and (c) persistence of the fiber within the lung. The effects of dose and dimension have been well-characterized from such studies and as a result are fairly well known in regard to human disease potential. The dose is obviously a product of the environment in which the fiber is used and the manner in which it is used. The dimension and persistence of the fiber within the lung, on the other hand, are functions of the manner in which the fiber is formed and of its chemical composition. general, the smaller the fiber the more likely that it will become embedded in lung tissue when inhaled, thus increasing the danger of respiratory disease.

Although less is known about the link between persistence of the fiber within the lung and respiratory disease, increasing attention is being focused on this Biological persistence aspect of the health issue. refers to the length of time a fiber endures as an entity within the body. The physiochemical concept that most closely relates to persistence and is perhaps more easily quantified is that of "durability" - specifically, the chemical solubility (or resistance to solubility) of fibers in body fluids and the tendency of such fibers to maintain physical integrity within such an environment. In general, the less durable a fiber is, the less will be the potential health risk associated with the inhalation of that fiber. measuring the chemical durability of a fiber in body fluids is to measure its durability in physiological

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saline solutions. This can be done by quantifying the rate of extraction of a chemical component of the fiber such as silicon into the physiological saline solution over a certain period of time.

5 Thus, as can be easily concluded from the foregoing discussion, conventional mineral wool fibers have several serious drawbacks. However, even the alternatives to mineral wools have problems. example, as mentioned earlier glass fibers have a fire resistance problem and whereas the refractory ceramic 10 fibers have been gaining increasing use in recent years as an alternative to mineral wool fibers because of their ultra-high temperature resistance and superior ability to pass all fire rating tests, their use is limited by the fact that they are relatively expensive 15 and have a relatively high chemical durability in physiological saline solutions as well.

In conclusion, there is a great need in the industry for low cost, friendly feeling low bulk density inorganic fibers which have good fire resistance properties as measured by their ability to pass the ASTM E-119 two hour fire test. Additionally, there is a tremendous demand for fibers which have especially low durabilities in physiological saline solutions. What would be particularly advantageous to the industry would be fibers with combinations of the above mentioned sought after properties. Also, advantageous would be fibers which also have excellent refractory properties as well, e.g. high continuous service temperatures.

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SUMMARY OF THE INVENTION

In one embodiment of the present invention, there are provided inorganic fibers having a silicon extraction of greater than about 0.02 wt% Si/day in physiological saline solutions and a composition consisting essentially of about 0-10 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 35-70 wt% SiO_3 ; 0-50 wt% MgO; and CaO.

In another embodiment of the present invention, there are provided inorganic fibers which have a 5 hour silicon extraction in physiological saline solutions of at least about 10 ppm. These fibers can broadly have compositions consisting essentially of the following ingredients at the indicated weight percentage levels:

0-1.5 wt% of either Al_3O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 40--70 wt% SiO_2 ; 0-50 wt% MgO; and CaO

1.5-3 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 40-66 wt% SiO_2 ; 0-50 wt% MgO; and CaO

3-4 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 40-64 wt% SiO_2 ; 0-50 wt% MgO; and CaO

4-6 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 40-59 wt% SiO_2 ; 0-25 wt% MgO; and CaO

6-8 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 35-54 wt% SiO_2 ; 0-25 wt% MgO; and CaO

8-10 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 35-45 wt% SiO_2 ; 0-20 wt% MgO; and CaO

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In a preferred embodiment, inventive fibers with 5 hour silicon extractions of greater than about 20 ppm and most preferably greater than about 50 ppm are provided.

5 In another embodiment of the present invention there are provided inorganic fibers having a diameter of less than 3.5 microns and which pass the ASTM E-119 two hour fire test when processed into a fiber blanket having a bulk density in the range of about 1.5 to 3 pcf; and having a composition consisting essentially of 10 0-10 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron about: oxides, or mixtures thereof; 58-70 wt% SiO_2 ; 0-21 wt% MgO; 0-2 wt% alkali metal oxides; and CaO and wherein the amount of alumina + zirconia is less than 6 wt% and the amount of iron oxides or alumina + iron oxides is 15 less than 2 wt%. Preferably, the inventive fibers in this embodiment may have compositions consisting essentially of about:

0-1.5 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 58.5-70 wt% SiO_2 ; 0-21 wt% MgO; 0-2 wt% alkali metal oxides; and CaO

greater than 1.5 wt% up to and including 3 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 58.5-66 wt% SiO_2 ; 0-21 wt% MgO; 0-2 wt% alkali metal oxides; and CaO

greater than 3 wt% up to and including 4 wt% of either Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides, or mixtures thereof; 58-63 wt% SiO_2 ; 0-8 wt% MgO; 0-2 wt% alkali metal oxides; and CaO

greater than 4 wt% up to and including 6 wt% of either Al₂O₃, ZrO₂, TiO₂, B₂O₃, iron oxides, or mixtures thereof; 58-59 wt% SiO₂; 0-7 wt% MgO; 0-2% alkali metal oxides; and CaO.

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As discussed herein earlier, there has been a demand in the industry for inorganic fibers with an excellent fire rating at low bulk densities and fibers with especially low chemical durabilities in physiological saline solutions. Therefore, each category of inventive fibers should fulfill a real need in the industry and should be available for applications where heretofore low cost, mineral wool type fibers have not been available. What is particularly advantageous about the present invention is the fact that fibers are provided where a special demand exists, i.e. applications in the industry where fibers with both an excellent fire rating and an especially low durability in physiological saline solutions are in demand.

Other features and aspects, as well as the various benefits and advantages, of the present invention will be made clear in the more detailed description which follows.

DETAILED DESCRIPTION OF THE INVENTION

The inventive fiber compositions of the present invention can be made from either pure metal oxides or less pure raw materials which contain the desired metal oxides. Table 1 herein gives an analysis of some of the various raw materials which can be used to make inventive fiber compositions. Physical variables of the raw materials such as particle size may be chosen on the basis of cost, handleability, and similar considerations.

Except for melting, the inventive fibers are formed in conventional inorganic fiber forming equipment

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and by using standard inorganic fiber forming techniques as known to those skilled in the art. Preferably, production will entail electric furnace melting rather than cupola melting since electric melting keeps molten oxides of either pure or less pure raw materials more fully oxidized thereby producing longer fibers and stronger products. The various pure oxides or less pure raw materials are granulated to a size commonly used for electric melting or they may be purchased already so granulated.

The granulated raw materials are then mixed together and fed to an electric furnace where they are melted by electric resistance melting with electrodes preferably positioned according to the teachings of U.S. Patent No. 4,351,054. Melt formation can be either continuous or batchwise although the former is preferred. The molten mixture of oxides is then fiberized as disclosed in U.S. Patent No. 4,238,213.

While the fiberization techniques taught in U.S. 4,238,213 are preferred for making the inventive fibers, other conventional methods may be employed such as sol-gel processes and extrusion through holes in precious metal alloy baskets.

The fibers so formed will have lengths in the range of from about 0.5 to 20 cm and diameters in the range of from about 0.05 to 10 microns with the average fiber diameter being in the range of about 1.5 to 3.5 microns. Table 2 shows the average fiber diameter (measured microscopically) and the unfiberized shot content of various inventive fibers. As may be seen, the average microscopic fiber diameter was 2.3 microns and the average unfiberized shot content was 27%.

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For purposes of comparison, conventional mineral wool fibers were also tested with the results being given in Table 2 as numbers 226 and 229. These conventional fibers averaged 4.7 microns (measured microscopically) in diameter and had an average 40 wt% shot content. The continuous service temperature ranged from 1370°F to 1490°F, averaging 1420°F.

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Table 3 contains an extensive chemical analysis of a number of inventive fibers. Because of the large number of fiber samples containing alumina additives made to the base calcium oxide/magnesia/silicasystem, only the average analysis of the minor constituent of these fibers are given in Table 3. The silica, alumina, magnesia, and calcium oxide contents for these fibers are given in Table 4.

As used herein, the "service temperature" of an inorganic fiber is determined by two parameters. The first is the obvious condition that the fiber must not soften or sinter at the temperature specified. this criterion which precludes the use of glass fibers at temperatures about 800°F to 1000°F (425° to 540°C). Additionally, a felt or blanket made from the fibers must not have excessive shrinkage when soaking at its service temperature. "Excess shrinkage" is usually defined to be a maximum of 5% linear or bulk shrinkage after prolonged exposure (usually for 24 hours) at the service temperature. Shrinkage of mats or blankets used as furnace liners and the like is of course a critical feature, for when the mats or blankets shrink they open fissures between them through which the heat can flow, thus defeating the purpose of the insulation. fiber rated as a "1500°F (815°C) fiber" would be defined

as one which does not soften or sinter and which has acceptable shrinkage at that temperature, but which begins to suffer in one or more of the standard parameters at temperatures above 1500°F (815°C).

5 The service temperatures for a representative number of fibers in the inventive compositional range are listed in Table 2. The continuous service temperature for constant silica/magnesia/calcium oxide ratios are given in Table 6. As may be seen in all cases, the lower the alumina content of the fiber, the higher the 10 service temperature will be, with the highest service temperature being at zero percent alumina for alumina contents less than 30%. Thus to attain the most desired properties of the inventive fiber it is not possible to accept any of the alumina contents resulting from 15 melting the traditional mineral wool raw materials. Rather, various amounts of sufficiently pure oxides will be required to dilute the alumina contents to the desired low levels. To attain fibers of the highest service temperatures, only pure raw materials with 20 essentially no significant amounts of alumina must be used.

A series of inventive fibers were also tested for their silicon extraction in a saline solution a saccording to the following procedure:

A buffered model physiological saline solution was prepared by adding to 6 liters of distilled water the following ingredients at the indicated concentrations:

Ingredient Concentration, q/1

MgCl₂6H₂O 0.160

NaCl 6.171

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	KC1	0.311
	Na ₂ HPO ₂	0.149
	Na ₂ SO.	0.079
	CaCl ₂ 2H ₂ O	0.060
5	NaHCO ₂	1.942
	$NaC_2H_3O_2$	1.066

Before testing, this solution was buffered to a pH of 7.6 by bubbling with a gaseous mixture of 5% $CO_2/95\%N_2$.

10 One half (1/2) gram of each sample of fiber listed in Table 3 was then placed into separate closed, plastic bottles along with 50 cc of the prepared physiological saline solution and put into an ultrasonic bath for 5 hours. The ultrasonic vibration application was adjusted to give a temperature of 104°F at the end of 15 the 5 hour period. At the end of the test period, the saline solution was filtered and the solution chemically analyzed for silicon content. The silicon concentration in the saline solution was taken to be a measure of the amount of fiber which solubilized during the 5 hour test 20 The CaO and MgO contents of the fiber were similarly solubilized.

One of the inventive fibers was tested for silicon extraction in a physiological saline solution for periods of up to 6 months. Results were as follows:

		Steady State	Total	Comments On
	Silicon	Silicon Extraction	Amphoteric	Fiber Residue
Fiber	Extraction	Rate For $0.20 \text{ m}^2/\text{g}$	Oxides in	After 6
Number	in 6 Months	Surface Area, \$ Si/day	Fiber	Months
29 (inventive)	%96	0.16%	1.0%	carbonate hydroxyl
•				apatite fiber,
				disintegrated into
				small particles
	ć	900	ð O	slight fine grained
137 (non-	\$P	\$6.40.0	•	4
inventive)				Ilbers with
				uniform corrosion
			,	
235 (non-	48	0.012%	25.6%	no fiber
inventive)				corrosion;
				some surface
				deposition

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Categorization of oxides melts according to scales of acidity or basicity has been well known for many years. (See "A Scale of Acidity and Basicity in Glass," Glass Industry, February 1948, pp 73-74.) have now found that by strictly controlling the compositions of the oxide melts according to the acidic or basicity behavior of the respective oxides, fibers can be made which are surprisingly soluble in saline solu-Increasing the content of silica, alumina, and the amphoteric oxides in the fiber increases the acid ratio of the fiber composition. This tends to stabilize the system against silicon extraction by weak solutions as a result of relative changes in the interatomic bonding forces and extension of the silica network. Other amphoteric oxides besides alumina will have an alumina equivalency with respect to extraction by saline solutions. The amphoteric oxides zirconia and titania appear to have an alumina equivalency of close to 1 to We have found that in general for desired high saline solubility the amount of total amphoteric oxides must be kept below about 10% depending upon the amount of silica present. On the other hand, with the exception of iron and manganese oxides, the basic oxides can vary widely since their alumina equivalency is small. However, while iron and manganese oxides are generally considered to be basic in nature, their behavior with respect to saline solubility more closely relate to the amphoteric oxides, thus the amounts of iron and manganese oxides must be similarly limited.

Many of the fibers were tested for their fire resistance according to the following simulated fire rating test procedure:

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For screening test purposes, a small furnace was constructed using an electrically heated flat-plate element at the back of the heat source. A 6 inch x 6 inch x 2 inch thick sample of 1 3/4 to 6 1/2 pcf density of each formulated fiber was mounted parallel with the element and 1 inch from it. Thermocouples were then positioned at the center of the fiber sample surfaces. A computer was used to control power via a simple on-off relay system to the heating element. The position of the relay was based on the reading of the thermocouple on the sample surface nearest the element and the programmed fire test heat-up schedule.

The furnace was heated so as to follow a standard ASTM E-119 time/temperature curve for the 2-hour test period. In the test utilized herein, failure of the fiber is considered to occur when the furnace is unable to maintain the standard temperature per ASTM E-119 because the fiber insulation has sintered sufficiently to allow heat to escape through the fiber layer.

The results of the testing of the fibers for 20 saline solubility and the two hour ASTM E-119 fire test are given in Table 4 for the fibers made with alumina addition and in Table 5 for the remaining fibers to which other oxidic constituents were added. additions included: B_2O_3 , P_2O_5 , TiO_2 , ZrO_2 , Fe_2O_3 + MnO, 25 La2O, Cr2O, and Na2O. For glass fibers within the scope of the invention to function in an ASTM E-119 fire test, i.e. to withstand the rising temperatures of a simulated fire which can reach 1850°F in two hours, it is necessary that they convert from an amorphous condition to a 30 beneficial pseudo crystalline state during heat-up. The inventive fibers do this but can be assisted in this function by the inclusion of suitable crystal nucleating

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agents. Such agents may include TiO_2 , ZrO_2 , platinum, Cr_2O_2 , P_2O_5 , and others. Such additions are within the scope-of this invention.

TABLE 1 RAW MATERIALS USED

			Pure Raw Materials	als		
	Silica Sand	Quick . <u>Lime</u>	Calcined <u>Dolomite</u>	Aluminum Oxide	Magnesium Oxide	
ACIDIC OXIDES						
Sio2	0.66	0.34	0.50	0.02	٥.	
AMPHOTERIC OXIDES	IDES				•	
\mathtt{TiO}_2	nil	nil	nil	0.002	nil	
A1203	0.30	0.26	0.50	98.8	0.1	
BASIC OXIDES) , ,	
Fe_2O_3	0.30	0.05	0.15	0.02	0.7	-1
MnO	!	1	!	ļ	1	5 –
Мдо	0.02	0.14	40.0	nil	96.3	
CaO	0.03	97.75	57.0	0.01	2.0	
Na ₂ O	0.04	0.02	0.01	0.30	0.02	
K ₂ O	0.01	0.01	nil	0.01	0.01	
MISCELLANEOUS						
SO,	!	į	0.4	1	!	
ູ່ເຂ	i	!	!	:	1	
ပ	;	:	;	!	_ i_	
IOI	0.2	0.7	3.0	0.20	1.8	
TOTAL	99.90	99.27	101.56	99.36	101:33	

TABLE 1
RAW MATERIALS USED (continued)

		Less Pure Raw Materials		
	Kaolin	Blast Furnace Slag	Nepheline Syenite	Talc
CIDIC OXIDES SiO ₂ MPHOTERIC OXIDES	50.5	35.16	61.3	61.2
Tio	1.61	0.62	0.003	nil
A1203	43.6	12.88	23.4	-17- L.0
Fe,O,	08.0	0.20	0.07	0.85
Mno	1	0.62	: 	† •
MgO	0.01	16.06	0.05	31.7
CaO	0.04	32.94	0.58	0.19
Na,O	90.0	. 0.45	9.60	1 1
K ₂ 0	0.02	0.25	4.50	
ITSCELLANEOUS				-
So	1	0.28	1	!
າແນ	!	1.03	: 1	1
ပ	i i	0.30	1 1	1 1
ĪO	2.90	1 1	0.62	5.0
FOTAL	99.54	100.79	100.12	0.00

Calcined Dolomite: Ohio Lime NO. 16 Burnt Dolomitic Lime Quick Lime: Mississippi Lime - Pulverized Quick Lime Ottawa Silica - Sil-co-Sil Grade 295 Reynolds Calcined Alumina, RC-23 Magnesium Oxide: Baymag 56 Feed Grade Aluminum Oxide: Silica Sand:

Kaolin: American Cyanamide Andersonville Kaolin Blast Furnace Slag: Calumite Morrisville Slag Nepheline Syenite: Indusmin Grad A400 Talc: Pfizer Grade MP4426

Additives:

55.5% B₂03 Soda Ash: 58.3% Na₂0 Boric Acid:

98.5% Iron Oxides Magnetite Iron Concentrates: Zircon:

66.2% Zro2

Chromium Oxide: 99.5% Cr₂0₃ Titanium Dioxide: 99% TiO $_2$ Manganese Oxide: 99% MnO₂

Moly Corp. Lanthanum Carbonate:

TABLE 3 COMPOSITION OF FIBERS

						AMERICANTER OXIDES	DES	1
		ACIDIC OX	KIDES	0110				SUB
TEST NO.	B,0,	<u>S10</u> 2	P_205	TOTAL	T102	$\underline{A1}_2\underline{o}_3$	$\overline{2}\overline{\text{LO}}_{2}$	TOTAL
	rton of	composition of Fibers With	Al,0, addi	Al, O, additions (minor	constitue	constituents only)		
TEORINO	17 110 71		7					0.02
1 to	00.00	ì	00.0	!	0.01	:	† • • • •	1 0 1- 0 1
	l I	1	Į	;	!	i i		
Composi	tion of	Composition of Fibers with	B203 additions	ions		(1	90.0
164	0.32	64.8		65.12	! 1	90.0		
י אַ	0.52	63.9	!	64.42	1	1.20	! !	02:1
601	0.64	64.6	!	65.24	;	90.0	1	90.0
167	0.82	64.5	\$ 1	65.32	!	90.0	! !	90.0
168	1.33	64.1	!	65.43	l l	90 * 0	; ;	90.0
169	1.37	64.1	!	65.47	!	0.00	1	90.0
170	2.22	63.6	†	65.82	1	90.0	: :	90.0
171	8.41	59.6	!	68.01	1	0.0	l	
Compos	Composition of	Fibers with	1 P ₂ O ₅ additions	tions	•	or c	0.04	0.48
2	!	49.6	6.05	55.65	0.06	0.38		
Compos	Composition of	Fibers with	h Tio2 additions		6	4 1 4	1	51.4
173	}	48.6	!	48.6	10.0	• •		
Compos	ition of	Composition of Fibers with 2r02 additions	h Zro2 addi		5	88	0.21	1.10
174	i	63.5	1 1	63.5	10.	0.33	0.40	0.73
175	!	59.2	!	59.2	i 1	0.31	0.42	0.73
176	!	59.5	!!	59.5	l I			

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TABLE 3 COMPOSITION OF FIBERS (CONTINUED)

		TABLE 3 COMPOSITION OF FIBER	TABLE 3 OF FIBERS (continued)	
		-	MISCELLANEOUS	-
ST.	<u>so</u> 3	Misc.	SUB <u>TOTAL</u>	TOTAL
omposition of	Fibers with Al ₂ 0	omposition of Fibers with Al203 additions (minor constituents only)	constituents only)	
4	.05/	.02	/10.	.14
}	.20		.22	.44
omposition of		additions		
	. ;	:	1 1	100.48
י ע ע	!	t 1	!	100.42
, y	!	1		100.5
9 7	I I	:	!!	100.58
/0	. 1	!	i i	100.39
80			! !	100.43
69		t 1		100 48
70	!	ţ	1 1	
17.	# I	ŧ .	!	100.07
composition of Fiber	f Fibers with P ₂ 0	rs with P ₂ O ₅ additions		1
7	ļ	0.02	0.02	99.73
Composition o	composition of Fibers with TiO2 additions	2_additions		
173	;		i 1	100.0
Composition o	Composition of Fibers with ZrO2 additions	2_additions		
174	;	1 1	\$ 1	100.52
175	!		!	יים קיים קיים
176	1 1	1 1	i 1	67.66

TABLE 3 COMPOSITION OF FIBERS

	SUB	TOTAL	-	-	0.84	06.0	0.93] . B.B.) <u> </u>) (2.89	2.69	2.95	3.53) (3.68	3.65	3.62	3.50	3 7 6	<u>0</u> (3.73	4.25	4.34	7.87
200	A L DES	$\overline{z}\underline{r}\underline{o}_2$			05.0	0.54	0.58	0.58	0.83		. o	2.31	2.65	3.11	ר ר	3.12	3.27	3.30	3.30	3, 16) (3.3/	3.67	3.69	4.50
AMDHOTEDIO OVIDE	PHENOTERIC O	A_{203}			0.34	0.36	0.35	1.29	0.32	2.03		0.38	0.30	0.42	0.56		0.38	0.32	0.20	0.39	96 0	0.0	0.58	0.65	3,35
		<u>110</u> 2			;	!	!	.01	;	.02		! !	1	!	!		!	į	;	i	;		i i	;	.02
	SUB		ZrO2 additions (Cont.)	50 7	7.00	0.09	59.2	54.3	59.2	46.85	59.4		39.05	57.96	57.80	. 65	00.00	56.88	57.7	58.19	57.86	λ Α		58.4	56.65
XIDES	P.02	5-7-	Zro2 add1	. ;	ļ	ļ [.]	!	!	!	1	-	!		j I	!!	;		! !	!	!	!	!	ļ	! !	;
ACIDIC O	<u>S10,</u>		campostrion or ribers with	59.7	0.09	, pr	2 · · ·	0.4°.3	23.5	46.85	59.4	59,05	57 96	יים היים היים	8./6	59.05	56 88		7.76	58.19	57.86	58.6	58.4	, L	56.65
	B ₂ O ₃	+ + +	JO UOTA	!	;	;	į	1	!	! !	!	1	;	!		!	!	ļ	i	!	ŀ	. !	!	!	
TP C.T	NO.	, accumos		177	80	179	180	ואנ	183	707	182(a)	183	184	185) (186	187	188	081	0 0	061	:91	.92	66)

TABLE 3
COMPOSITION OF FIBERS (continued).

	SUB TOTAL		39.16	38.78	37.98	43.12	37.73	49.98	36.96	38.07	38.72	38.14	39.51	40,45	39.0	38.65	38.88	36.22	35.79	35.36
	<u>K</u> 20		1	!	;	.02	; 	.01	 	.01	;	l l	1	!	!	!	i	1		.01
	Na ₂ 0		!	1	1 1	. 04	1	.05		.03	1	i	!	1	1	ľ	;	1	1	.05
	<u>BaO</u>		!	ŀ	;	.01	t I	.03	1	00.	1	!	!	 	t I	!	1	1	-	00.
S	<u>Ca0</u>	7	38.7	.38.3	37.0	32.75	36.6	29.5	34.9	34.84	35.17	34.4	36.94	36.45	36.0	35.39	35.66	33.5	33.2	31.9
BASIC OXIDES	<u>Li20</u>	S (Cont	1 1	;	;	1	!	1	l I	!	1	ŀ	!	! !	1	!	!	i i	!	-
BASI	<u>O</u>	ditions	0.46	0.48	0.98	10.20	1.13	20.6	2.06	3.08	3.55	3.74	2.57	4.00	3.00	3.26	3.22	2.72	2.59	3.35
	$\frac{C_{L}}{2}$	ZrO2 additions (Cont.)	1	1	ij	1	!	: !	ţ	.05		!!	1	E E	: :	i	1 1	ŀ	!	00.
	<u>La 203</u>	bers with	1	!	¦	!	1	!		!	!	ţ	1			i	;	!	1	}
	Mno	of Fibe	! !	1	;	.01	1 1	.01	1	00.	!	1	1	1	!	1	1 1	ľ	!	00.
	Fe03	Composition of Fi	1						1					!			;	!	ľ	. 05
	rest 10.	Compos	177	80	179	081	181	182	182(a)	183	184	581	981	281	881	681	061	161	192	193

TABLE 3 COMPOSITION OF FIBERS (continued)

86 87 88 100.95 89 100.20 90 90 91 91 91 92 93 93 93 94 95 96 96 96 96
.0.

TABLE 3 COMPOSITION OF FIBERS

		ACIDIC OXIDES	IDES			AMPHOTERIC OXIDES	DES	
EST 0.	B_2O_3	<u>Si0</u> 2	P205	SUB	TiO2	A1203	<u>2r0</u> 2	SUB
omposi	tion of F	omposition of Fibers with	FeO, and N	Mno additions				
94	!	64.9	, ¦	64.9	{	90.0	;	0.06
95	i	49.8	;	49.8	.01	18.0	.01	18.02
96	ľ	50.4	1	50.4	.03	7.45	.01	7.49
97	i	64.34	!	64.34	1	90.0	1	0.06
86	i	63.70	;	63.70	i	1.20	1	1.20
66	i	63.54	i	63.54	1 1	1.20	;	1.20
00	ł	38.9	;	38.9	.01	6.70	.01	6.72
01	;	64.3	i	64.3	1 1	90.0	:	0.06
03	1	44.6	!	44.6	.01	0.92	.01	0.94
03	!	63.3	ļ	63.3	ì	1.15	!	1.15
04	i	63.6	;	63.6	1	90.0	:	90.0
05	1	43.8	;	43.8	.01	15.26	.01	15.28
90	!	62.3	ł	62.3	!	1.20	;	1.20
07	:	63.3	ļ	63.3	.	90.0	;	0.0
80	1	43.9	;	43.9	.01	14.3	.01	14.32
60	1	62.0	i	62.0	!	90.0	¦	0.06
10	ļ	0.09	;	0.09	;	2.0	i	2.0
11	i i	0.09	1	0.09	į	;	1 1	

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TABLE 3
COMPOSITION OF FIBERS (continued)

Troub.					BASI	BASIC OXIDES	S				
NO.	FeO ₃	Mno	La_2 03	$\frac{Cr}{2}$	Mgo	1.120	<u>Ca0</u>	<u>Ba0</u>	Na ₂ 0	K ₂ O	SUB TOTAL
Compo	Composition of Fi	_	bers with	FeO, and Mno	od Mno	additions	ns		1	3	
194	0.06	1	1	, ¦	8.72		26.6	!	!	;	15 10
195	. 22		;	!	0.2	;	31.5	ł	1	j	00.00
196	. 48	.04	;	;	15.2	;	26.2	ļ	0.7	ע	31.32
197	.50		!	!	7.80	ļ	26.4	!	· ;	. ! . !	46.04
198	69.	1	;	1	7.73	1	25.30	1	1	1	73.72
199	.72	i	. !	!	7.70	;	25.04	;	!	!	33.46
200	.80	;	1.	ļ	16.1	1	37.5	!	1	!	54.40
201	96.	;	!	1	8.6	;	26.4	i	ł	!	35.96
202	1.02		ľ	!	18.1	i	32.8	!	i	!	52:55
203	1.61		!	!	7.98	;	25.4	!	1	:	96.10
204	1.92		!	1	8.6	1	26.1	.	!		מייים מיי
205	2.90	.04	}	.14	22.7	;	15.05	!	.10		30.02
206	3.05		!	!	8.0	1	25.0	1	}	: ;	36.05
207	3.45	!	ţ	Į,	8.0	i	25.5	;	;	!	36.95
208	3.50	ļ	1	!	24.4	:	13.7	!	:	;	41.6
509	4.81	i	:	i i	8.0	!	25.5	1	;	1	38.31
210	į	8.0	!	!	30.0	1	:	:	1		38.0
211	;	20.0	!	I I	20.0	!	-	ţ	!	i	40.0

COMPOSITION OF FIBERS (continued).

MISCELLANEOUS	SUB TOTAL		100.34	.07	.07	99.1	98.62	98.20	.07	100.32	97.46	99.44	100.28	.13	99.55	100.31	99.82	100.37	100.0
	TEST SO ₃ Misc.	position of				197					202		204			207	208		010

TABLE 3 COMPOSITION OF FIBERS

	SUB	TOTAL	-	90.0	90.0	90.0) (90.0		0.51	~		90.0	90:0	90,0		1:20	90.0	90.0	90.0	0	00.	90.0
00017	210	2		1	1	!	!	•		0.01			! !	!	;	1		:	;	i	i		! :
AMPHOTEBTC OVIDES	Alo	5-7		90.0	90.0	90.0	90.0		9	U. 4		90 0	•	90.0	90.0	1.20		0.00	90.0	90.0	0.06	30.0	
	Tio	7		1 1	!	1	ļ			70.0		!		1	!	1	;		!	!	;	!	
	SUB	ditions	1 88	• • •	8·/c	57.5	56.9	tions	62.6	ions.		64.7	¥ 7.9		64.4	63.5	64.3		7.40	64.0	63.0	60.3	
OXIDES	P ₂ 05	La,	· · ·	ļ		ľ	!	Cr,0, additions) ! !	Na O additions	7	!	;		J I	!	i	!		1	!	l i	
ACIDIC OXIDES	<u>S10</u> 2	Composition of Fibers with	58.1	57.8	, L	, i	26.9	composition of Fibers with	62.6	Composition of Fibers with		64.7	64.5	7 79	r •	63.5	64.3	64.2	· · · · ·) (0.50	60.3	
	B ₂ O ₃	sition of	!	;	;	1		Sition of	!	sition of	1	i	;	;		!	!	;	;	!	1	;	
T.S.R.T.	NO.	Compo	;	213	214	215		COMPO	2216	Compo	<u>-</u>	•	i 218	1219	200	9 6	777	222	223	224		522	

TABLE 3 (COMPOSITION OF FIBERS (CONTINUED)

GIF	SUB CaO BaO Na $_2$ O $_{ m K_2}$ O TOTAL		36.71 41.47	36.53 41.82	36.3 41.72	36.0 41.58		34.10 0.00 0.03 0.01 36.61		26.6 0.28 35.58	26.5 0:45 35.65	26.5 0.71 35.80	26.1 0.87 35.70	26.2 0.93 35.63	26.4 1.11 36.11	26.3 1.40 36.3	25.9 2.60 37.0	74.74
BASIC OXIDES	1.4203 Cr.203 Mg0 1.120	ers with La ₂ 0 ₃ additions	0.00 4.60	0.56 4.58	0.72 , 4.55	0.92 4.51	ers with Cr ₂ 0 ₃ additions	0.09 2.30	ers with Na, O additions	7.8	7.8 B.7	8.6	8.5	8.5	8.6	8 8 6	8.5	
	TEST FeO3 MnO	Composition of Fiber	0.16	0.15	214 0.15		Composition of Fibe	216 0.08 .00	sition of Fib		1		000	(66	222	223	224	

TABLE 3
COMPOSITION OF FIBERS (continued)

į		TOTAL			99.63	89.66	99.28	98.54		99.72		100.34	100.21	100.26	100.40	66.66	100.37	100.36	100.06	100.1
	MISCELLANEOUS	SUB TOTAL		!		i	1	;		;		! !	;	•	!	!	•	!	:	:
•		Misc.	Composition of Fibers with La ₂ O ₃ additions	1	!	;	!	Composition of Fibers with Cr.O. additions		th Na,O additions			!	ľ	1	:	:	!	!	!
		$\frac{50}{3}$	tion of Fibers wi		;	i	. 1	tion of Fibers wit		Composition of Fibers with Na,0 additions	ļ	;	i	!		!	} }	}	i i	! !
	TEST	NO.	Compost	:	213	214	215	Composi	216	Compost	17	218	219	220	22.1	222	223	224	225)

TABLE 3

COMPOSITION OF FIBERS

Name Page Page			ACIDIC OXIDES	(IDES			AMPHOTERIC OXIDES	OXIDES	
SiQ2 E265 TOTAL TIO2 A1203 ZEO2 Conventional Mineral Mools 40.0 0.37 9.1 0.03 40.0 - 40.0 0.37 9.1 0.03 39.9 0.02 39.92 1.11 12.85 0.03 37.65 0.84 38.49 2.35 9.85 0.04 41.75 0.12 41.87 1.07 16.0 0.04 37.1 - 41.67 1.07 16.0 0.03 37.1 - 31.0 - 47.5 0.02 37.1 - 47.5 0.02 - 6.02 - 50.0 - 31.0 - 40.0 - 6.03 - 50.0 - 50.0 - 46.0 - 6.03 - - 6.03 - - 6.03 - - 6.03 - - - - - - - -	ŀ				SUB				รูกธ
Conventional Mineral Mools 40.0 - 40.0 0.37 9.1 0.03 39.9 0.02 39.92 1.11 12.85 0.03 37.65 0.84 38.49 2.35 9.85 0.04 41.75 0.12 41.87 1.07 16.0 0.04 41.75 0.12 41.87 1.07 16.0 0.03 31.0 - 31.0 - 47.5 0.02 31.1 - 31.0 - 47.5 0.02 37.1 - 50.0 - 40.0 - 69.2 - 50.0 - 50.0 - 40.0 - 60.0 - 54.0 - 50.0 - 40.0 - 60.0 - 60.0 - 60.0 - 60.0 - 60.0 - 60.0 - 60.0 - 60.0 - 70.0 - 70.0 -		2 <u>0</u> 3	<u>S10</u> 2	P_20_5	TOT'AL	$\frac{\text{TiO}_2}{2}$	$A1_20_3$	$\frac{2 \text{ CO}}{2}$	TOTAL
40.0 - 40.0 0.37 9.1 0.03 39.9 0.02 39.92 1.11 12.85 0.03 37.65 0.84 38.49 2.35 9.85 0.03 41.75 0.12 41.87 1.07 16.0 0.04 41.75 0.12 41.87 1.07 16.0 0.03 31.0 - 31.0 - 47.5 0.03 50.0 - 37.1 - 59.2 - 50.0 - 50.0 - 40.0 - 58.47 1.15 59.62 0.98 24.54 0.03 58.1 - 52.0 - 46.0 - 52.0 - 52.0 1.76 44.4 .23 52.0 - 52.0 1.76 44.4 .23 49.8 - 49.8 1.60 38.3 9.32 48.6 - 47.8 1.50 34.4 15.1 46.2 - 46.2 1.40 1.50 34.4	1		onvention		1 Wools				
39.9 0.02 39.92 1.11 12.85 0.03 37.65 0.84 38.49 2.35 9.85 0.04 41.75 0.12 41.87 1.07 16.0 0.04 of Refractory Fibers Fibers with less than 25% Basic Oxides 0.03 31.0 - 47.5 0.03 50.0 - 37.1 - 47.5 0.02 50.0 - 37.1 - 40.0 - - 69.2 - <t< td=""><td></td><td>ı</td><td>40.0</td><td>ı</td><td>40.0</td><td>0.37</td><td>9.1</td><td>0.03</td><td>9,50</td></t<>		ı	40.0	ı	40.0	0.37	9.1	0.03	9,50
37.65 0.84 38.49 2.35 9.85 0.04 41.75 0.12 41.87 1.07 16.0 0.03 of Refractory Fibers Fibers with less than 25\$ Basic Oxides 31.0 - 47.5 0.03 37.1 - 47.5 0.02 50.0 - 59.0 - 40.0 - 54.0 - 54.0 - 46.0 - 58.47 1.15 59.62 0.98 24.54 0.03 52.0 - 52.0 1.76 44.4 .23 52.0 - 52.0 1.76 38.3 9.32 49.8 - 49.8 1.50 38.3 9.32 47.8 - 47.8 1.50 34.4 15.1 46.2 - 46.2 2.4 1.51 46.2 - 47.8 1.50 34.4 15.1 46.2 - 46.2 2.8 1.40 20.7 28 - 46.5 1.50 31.0		t	39.9	0.02	39.92	1.11	12.85	0.03	13.99
of Refractory Fibers (Fibers with less than 25% Basic Oxides) 1.07 16.0 0.03 31.0 - 47.5 0.02 37.1 - 47.5 0.02 37.1 - 47.5 0.02 50.0 - 40.0 - 69.2 - 54.0 - 54.0 - 46.0 - 58.47 1.15 59.62 0.98 24.54 0.03 52.1 - 52.0 - 44.4 .23 52.0 - 52.0 1.76 44.4 .23 49.8 - 49.8 1.50 38.3 9.32 48.6 - 48.6 1.50 34.4 15.1 46.2 - 46.2 1.40 31.0 20.7 28 - 28 1 27.4 - 64.5 - 27.4 - 27.4 -		t	37.65	0.84	38.49	2.35	9.85	0.04	12,24
Of Refractory Fibers With less than 25% Basic Oxides 31.0 - 47.5 0.02 37.1 - 47.5 0.02 50.0 - 40.0 - 54.0 - 46.0 - 58.47 1.15 59.62 0.98 24.54 0.03 58.47 1.15 59.62 0.98 24.54 0.03 52.1 - 52.0 1.76 44.4 .23 49.8 - 49.8 1.71 42.2 2.93 48.6 - 48.6 1.60 38.3 9.32 46.2 - 47.8 1.50 34.4 15.1 46.2 - 28 1.40 31.0 20.7 28 - 28 19 50 3 64.5 - 64.5 - 27.4 -		4	41.75	0.12	41.87	1.07	16.0	0.03	17,10
of Refractory Fibers With less than 25% Basic Oxides 31.0 - 47.5 0.02 37.1 - 37.1 - 47.5 0.02 50.0 - 50.0 - 46.0 - 54.0 - 54.0 - 46.0 - 58.47 1.15 59.62 0.98 24.54 0.03 52.1 - 52.0 1.76 44.4 .23 49.8 - 49.8 1.71 42.2 2.93 48.6 - 47.8 1.50 38.3 9.32 46.2 - 46.2 1.40 31.0 20.7 28 - 28 1 1.40 31.0 20.7 64.5 - 64.5 - 27.4 -									_
31.0 - 47.5 0.02 37.1 - 37.1 - 69.2 - 50.0 - 50.0 - 40.0 - 54.0 - 54.0 - 46.0 - 58.47 1.15 59.62 0.98 24.54 0.03 52.1 - 52.1 1.76 44.4 .23 52.0 - 52.0 1.76 44.4 .23 49.8 - 49.8 1.60 38.3 9.32 48.6 - 48.6 1.55 36.2 12.3 46.2 - 46.2 1.50 34.4 15.1 28 - 28 19 50 3 64.5 - 27.4 - - - -	1	of	Refractory	Fibers	Fibers with 10	ess than 25% Basic	Oxides)		
- 37.1 - 59.2 - - 50.0 - 40.0 - - 54.0 - 46.0 - - 52.1 0.98 24.54 0.03 - 52.1 1.76 44.4 .23 - 49.8 1.71 42.2 2.93 - 48.6 1.60 38.3 9.32 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -			31.0		31.0		47.5	0.02	47.52
- 50.0 - 40.0 - - 54.0 - 46.0 - 7 1.15 59.62 0.98 24.54 0.03 - 52.1 1.76 44.4 .23 - 52.0 1.71 42.2 2.93 - 49.8 1.60 38.3 9.32 - 48.6 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		ı	37.1	1	37.1	1	59.2	ı	59.2
- 54.0 - 46.0 - 7 1.15 59.62 0.98 24.54 0.03 - 52.1 1.76 44.4 .23 - 49.8 1.71 42.2 2.93 - 48.6 1.60 38.3 9.32 - 47.8 1.55 36.2 12.3 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -	•	ı	50.0	ı	50.0		40.0	t	40.0
7 1.15 59.62 0.98 24.54 0.03 - 52.1 1.76 44.4 .23 - 49.8 1.60 38.3 9.32 - 48.6 1.55 36.2 12.3 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		ı	54.0	ι	54.0	1	46.0	ı	46.0
- 52.1 1.76 44.4 .23 - 52.0 1.71 42.2 2.93 - 49.8 1.60 38.3 9.32 - 48.6 1.55 36.2 12.3 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		ı	58.47	1.15	59.62	0.98	24.54	0.03	25.55
- 52.0 1.71 42.2 2.93 - 49.8 1.60 38.3 9.32 - 48.6 1.55 36.2 12.3 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		1	52.1	ı	52.1	1.76	44.4	. 23	46.39
- 49.8 1.60 38.3 9.32 - 48.6 1.55 36.2 12.3 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -			52.0	1	52.0	1.71	42.2	2.93	46.84
- 48.6 1.55 36.2 12.3 - 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		ı	49.8	1	49.8	1.60	38.3	9.32	49.22
- 47.8 1.50 34.4 15.1 - 46.2 1.40 31.0 20.7 - 28 19 50 3 - 64.5 - 27.4 -		1	48.6	t	48.6	1.55	36.2	12.3	50.05
.2 - 46.2 1.40 31.0 20.7 - 28 19 50 3 .5 - 64.5 - 27.4 -		1	47.8	ı	47.8	1.50	34.4	15.1	51.00
- 28 19 50 3 .5 - 64.5 - 27.4 -		ı	46.2	ı	46.2	1.40	31.0	20.7	53.10
.5 - 64.5 - 27.4 -		ı	28	ı	28	19	50	E	72
		1	64.5		64.5	1	27.4	1	27.4

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TABLE 3 (cont'd.)

	SUB SUB SC. TOTAL TOTAL	19 100.16 4 100.47 1 100.69	4 101.14	0	99.66	100	100	100.11	99.62	99.91	100.04	99.65	99.78	100.23	100	100.3
	SUB	0.69	0.64	1	1	ı	ı	0.71	ı	ı	ł	ı	i	ı	ı	ı
2	Misc.	0.59 0.07 0.19	0.08	ı	. 1	ı	1	0.24	ı	ı	ı		1	ı	ı	
	503	0.1 0.67 0.42	0.56	ı Q	ı	i	ı	0.47	ı	1	i	ı	ı	1	ı	1
	SUB	45.82).63 41.53 0 Basic Ovided	21.4	3.3	10.0	ı	14.23	1.13	1.07	1.02	1.00	. 98	0.93	1	8.4
31	K20	0.55 0.27 0.80	_	1	ı	ı	1 .	1.18	90.	90.	90.	90.	90.	90.	ı	ı
	Na ₂ 0	0.54	2.04 an 25	20.2	3.1	4.4			. 05	. 05	.05	. U.S		60.	1 .	ľ
	Bao	0.04	ss th	1	ı		, ,	0.54	1	1	t i	ı	l 1	١,	· •	
ES	<u>Ca0</u>	36.5 38.55 23.55	s (Fibers with less than 25%	1.2	0.5	9°6)	21.0	0.12	0.12	0.12	0.12	2 4 • 1	ı	
BASIC OXIDES	MGO L1 ₂ 0 neral Wools	0.01 0.01 0.01	bers	ı	<i>i</i>	l i	0.02	30.1	1	ı	1	1	ı	,	4	
BASI	Mineral	11.2 6.05 12.95 6.45	ers (Fi	ı	1 1		1.44		0.07	0.07	0.07	0.07	0.07	i	1	
	$\frac{Cr_2O_3}{ional}$	0.02 0.04 0.04	Composition of Refractory Fiber	1 1	, ,	1	0.00	ı	ı	1	ı	ı	ı	,	i	
	NO. FeO ₃ MnO La ₂ O ₃ Cr ₂ O ₃ Composition of Conventional 226 0.47 0.64	111	efracto		ı	i	1	ı	ı	ı	ı	ı	1	1	1	
	Mno n of (0.24 0.22 0.23	of R	l ı	ı	ı	0.02	ı	ı	ŧ	1	1	ı	1	1	
	FeO ₃ Ssitio	0.35 9.7 3.75	sition -	ı	i	1	3.70	.83	.77	.72	.70	.68	. 63	ı	ı	
TEST	NO. Compo	228	Compo 233	232	233	234	235	236	237	238	339	:40	:41	. 4 2	43	

	ADDITIONS	
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. .	TEST RESULTS ON FIBERS MADE WITH A	
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DATA			Saline	oral Extraction Thickness 2 Hour	ppm. Si Density				30	51	69 .	.39 70 2.0/1.87 F	47	.30 46) u	0	ı	59	80	.61 49 2.0 /1.91 F	61	74	, c	ט מ			
			İ		1		2.0/1.97	1	1	2.0/1.94	2.0/2.12	2.0/1.87		I	I	1	1.88/2.20	2.0 /1.97	2.0 /1.91	2.0 /1.91	2.0 /1.91	2.0 /1.91	2.0 /1 94	2 0 1/ 0 2	2.0 / 1.91	7.01	†	
	3	Jnou c	Saline	Extractio	. mdd		8 3	89	30	51	69 .	70	47	46	40		0	ı	59	80	49	61 .	74	, 8 .	0 C	א נ))	
WIND THE DATA			E	Total	Analytical	6	100.20	100.47	79.66	09.60	100.57	99.39	99.97	100.30	100.10	99.56	0 0	99.83	50.64 50.64	99.94	99.61	100.54	99.22	99.39	99.32	100.98	Failed	
			Basic Oxides	201102	MgO Total	10 8 46 0		36 5 44 1				0.23 42.75	7.39 42.59	17.6 42.2	6.84 41.94	3.95 41.1	6.2 41 05		4.33 41.33	4.79 40.59	0.31.41.21	26.3 41.7	5.36 40.46	0.27 40.57	5.6 40.1	6.2 41.7		
	I. WT&		Basi		<u>a0</u> -	1.5	ונכ	<u>ب</u>		· -	,	,	12.4	24.5	35.0	36.95	34.75				(35.0	40.2 (34.4	35.4	d = d **	
	COMPOSITION, WT&	Amphoteric	Oxides		<u>A1</u> 2 ² 3 <u>Total</u> <u>C</u> Amphoteric Oxides	0.35	0.42	1.02	0.10	0.42	0.24			1.05	1.11	0.94	0.78	0.05	1.10	3 0		٠. ت د . ت	0.11	0.07	0.53	0.43		
•	S	Amph	0	1.4		•	0.40	1.00	0.08	0.40	0.20	6	•	1.03	1.09	0.92	0.75	0.03	1.08	0.03	22.0		0.09	0.05	0.49	0.41	erizable	
		Acidic	Oxides	o i s		53.8	53.9	54.5	55.9	56.0	56.35	56.4		97.0	57.0	57.25	57.8	58.1	58.2	58.3	58.4		08.0	58.7	58.5	58.8	Not Fiberizable	
				Q.	0 t.	20	21	. 22	23	24	35	9;	ŗ	. :	∞	ō.	0	-	7	m	4	Ľ	n	9	7	m	11	

EXPERIMENTAL DATA

									,	- 3 5	-													
	<u>'rest</u>	2 Hour	Test*		a	a	<u>a</u>	<u>-</u>	2	2	<u>-</u>	Ŀ	í.	-	a,	d	4	2	-	<u>.</u>	<u>-</u>	<u>-</u>		
	E-119 Fire fest	Thickness	Density		2.0/1.86	2.0/1.97	2.0/1.90	2.5/1.4	2.0/1.95	2.0/1.92	2.0/1.90	2.0/1.89	2.0/1.88	2.0/1.91	2.0/2.01	2.0/1.98	2.0/1.95	2.0/1.91	2.0/1.89	2.0/1.95	2.0/1.94	2.0/1.93		- • •
5 Hour	Saline	Extraction	ppm. Si		29	49	89	47	. 09	61	77	73	51	70	30	47	45	41	59	45	36	51	56	
		Total	Analytical		99.45	99.21	100.09	101.11	99.94	100.11	99.87	99.95	100.8	6.66	100.86	100.55	100.78	100.58	99.30	99.97	100.68	76.99	100.17	= Failed
		ic Oxides	MgO Total		6.10 40.4	3.8 39.9	0.43 40.83	36.8 41.60	4.75 40.40	10.7 40.60	5.98 40.28	8.16 40.36	16.8 39.6	11.4 40.2	0.11 40.71	12.9 39.0	11.0 39.4	16.4 39.0	6.36 38.76	9.85 38.45	10.7 38.9	9.47 38.27	3. 39.10	Poor, F
WT\$		Basi	Ca0	les	34.2	35.9	40.3	4.7	35.55	29.8	34.2	32.1	22.5	28.7	40.5	25.8	28.1	22.3	32.3	28.5	27.9	28.7	36.	= d **
COMPOSITION	eric	Oxides	Total	Amphoteric Oxides	0.10	0.26	0.11	0.26	0.34	90.0	0.04	0.04	1.45	0.05	0.30	1.50	1.33	1.43	0.19	1.07	1.13	0.95	0.22	e U
COME	Amphoteric	0x1	$A1_20_3$		0.08	0.24	0.09	0.24	0.32	0.04	0.02	0.02	1.43	0.03	0.28	1.48	1.31	1.41	0.17	1.05	1.11	0.93	0.2	Not Fiberizable
	Acidic	Oxides	S10 ₂	1 1/2\$	58.9	59.0	59.1	59.2	59.15	59.4	59.5	59.5	59.6	59.6	59.8	59.9	59.9	60.09	60.3	60.4	60.5	60.7	8.09	
			NO.	0 to	39	40	41	42	43	44	45	46	47	48	20	51	52	53	54	55	56	57	58	ii ≉

		Test.	2 Hom:	TPS:		a	. a.	. 2.	à	ù	<u>a</u>	<u>a</u>	೭	ď	ď	ď	d	ſτ	c.		1 2.	<u>-</u>	Ŀ	
		E-119 Fire Test	Thickness	Density		2.0/1.97	2.0/1.88	2.0/1.92	2.0/1.82	2.0/1.95	2.0/1.96	2.0/1.91	2.0/2.01	2.0/1.88	2.0/1.88	2.0/1.99	2.0/1.91	2.0/1.88	2.0/2.00		2.0/1.87	2.0/1.91	2.0/1.88	
	5 Hour	Saline	Extraction	ppm. Si		65	76	99	64	46	19	12.	52	17	7	49	37	46	35	44	30	25	46	
EXPERIMENTAL DATA			Total	Analytical	-	89.66	99.81	99.63	06.66	79.69	99.92	100.06	99.29	86.66	99.07	99.17	99.58	99.94	99.68	99.80	99.80	99.78	99.84	F = Failed
EXPERIM			asic Oxides	MgO Total		5.19 37.89	5 37.3	6.64 37.04	7.70 37.30	5.28 36.48	2 35.5	9 35.0	5.79 34.29	8 34.7	2.60 33.67	4.83 33.53	6.68 34.18	1 33.32	6.50 34.0	5.21 33.91	8 33.8	7.88 33.78	1 33.23	= Poor,
	3,0	٠	Basic (9	.7 15.5	Э	.5	.1	.2 10.2	.0 10.9	4	.8 11.8	30.97 2.		4	12 30.1	4		9 11.8		.12 30.1	4 * P
	N, WT&			<u>Cao</u>	<u>kides</u>	32.	21.	30.	29	31	25	24	28.	22.8	30	28.6	27.	3.	27.	28.6	21.9	25.8	Э.	
	COMPOSITION	Amphoteric	0xides	Total	Amphoteric Oxides	0.04	0.06	0.04	0.05	0.04	1:27	1.51	1.15	1.43	1.25	1.49	0.05	1.17	0.03	0.04	0.05	0.05	1.17	
	COM	Ampho	XO	$\frac{A1}{2}$	1	0.02	0.04	0.02	0.03	0.02	1.25	1.49	1.13	1.41	1.23	1.47	0.03	1.15	0.01	0.02	0.03	0.03	1.15	Not Fiberizable
		Acidic	Oxides	$\frac{\text{Si0}_2}{\text{2}}$	0 to 1 1/2\$	61.7	62.4	62.5	62.5	63.1	63.1	63.5	63.8	63.8	64.1	64.1	65.3	65.4	9.59	65.8	62.9	65.9	65.4	dot Fibe
				NO.	0 to	59	. 09	61	62	63	?	S IR	ğ IT?	711 711	11	6 9	FE	171	72	73	74	75	. 9/	

		e Test	2 Hour	Testar	:	4 .		• •	:	<u>.</u>	ı .	<u>.</u> 1			4	к <u>(</u>	4 9	L	1 2	. 2	. =	. а	. 1	â	-
		mbigle	Thickness	_VEIISTEY_		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	2 0/2 03	(0.2/0.2	00 670 6	2.0/2.00	00 6/0 6	00.2/0.2			•	30 (70 6	2.0/1.88	2.0/1.05	2.071 99	2.0/1.82	2.0/1.87	2.0/2.06		2.0/1.98	
	5 Hour	Bytraction	pom si		50	78	84	· *	18	3.1	30	18			1	5.1	39	65	4 4	18	σ	25	11	29	
<u>EXPERIMENTAL DATA</u>		Total	Analytical		99.88	100.18	100.06	100.25	99.11	100.45	6.66	100.05			100.17	100.27	99.58	99.89	100.71	99.33	99.40	99.24	99.91	99.85	ed.
М ТИЗАХЗ		ides	Total		33.02	33.03	32.77	31.8	30.19		30.9	31.0			48.1	45.6	41.0	41.8	40.43	38.1	37.4	37.1	37.5	38.0	F = Failed
		Basic Oxides	MgO	nt.)	4.02 28.7	6.43 26.5	67 24.0	30.1	1.09	21.3	12.7	23.8			43.0	41.7	10.6	17.3	36.3	1.4	1.0	2.1	10.0	9.9	Pass,
64 E	1	Bas	<u>Ca0</u>	les (Co	4.02	6.43	8.67	1.6	29.0	10.2	18.1	7.2		des	5.0	3.8	30.3	24.4	3.83	36.6	36.3	34.9	27.4	28.0	= d *
COMPOSTATON	Amphoteric	Oxides	Total	Amphoteric Oxides (Cont.)	0.61	· 1	0.04	ı	0.27	ı	0.05			1/2% to 3% Amphoteric Oxides	2.02	2.03	2.43	1.84	2.03	2.28	2.95	2.69	2.56	1.70	
Ö	Ampho	0	$A1_{2}0_{3}$	- 1	0.59	ı	0.02	ľ	0.25	ı	0.03			Ampho	2.00	2.00	2.41	1.82	2.01	2.26	2.93	0.38	2.54	1.68	rizable
	Acidic	<u>Oxides</u>	Si0 ₂	0 1 1/2\$	66.1	67.1	67.2	68.4	9.89	68.8	68.89	0.69		/2% to 3%	50.0	52.6	56.1	56.2	58.1	58.9	29.0		59.8	60.1	Not Fiberizable
			NO.	0 to	77	78	79	80	81	8 6	n n	4 	~. .	1 1	82	98	87	88	83	06	91	95	93		 4

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		Test	2 Hour	Test. * *			<u>a</u>	24	2 ,	÷	a,	a,	<u>.</u>	a	a.	Ь	ď	ď		1	£.,	2.	
		E-119 Fire Test	Thickness	Density		2.0/2.04	2.0/1.87	2.0/1.91	2.0/1.93	2.0/1.90	2.0/1.91	2.0/1.96	2.0/1.87	2.0/1.94	2.0/1.95	ı	2.0/1.91	2.0/1.90		1	2.0/1.96	2.0/2.06	
	5 Hour	Saline	Extraction	ppm. Si		50	18	61	51	55	13	18	37	38	12	17	33	2		33	19	33	
EXPERIMENTAL DATA			Total	Analytical		100.18	100.04	100.03	99.01	99.28	99.02	99.66	99.05	99.11	100.4	100.57	99.73	99.47		99.65	69.66	100.93	Failed
EXPERIME			des	Total		37.7	36.4	36.9	34.3	34.4	34.1	35.1	33.4	33.3	34.3	33.15	32.5	30.9		46.18	45.74	41.89	li Cen
			ic Oxides	M90	ont.)	4.9	10.1	6.9	0.2	0.2	0.3	9.4	0.2	2.5	16.3	23.1	29.7	0.1		40.9	0.64	33.7	Pass,
	MT8		Bas	Ca0	9	32.7	26.2	29.9	34.0	34.1	33.8	25.6	33.1	30.7	17.7	9.74	2.7	30.7		4.98	45.0	7.89	11 Q * *
	COMPOSITION	Amphoteric	Oxides	<u>Total</u>	3% Amphoteric Oxides	2.23	2.19	1.68	2.86	2.83	2.77	1.81	2.56	1.86	1.85	2.17	1.58	1.82	Oxides	3.52.	3.60	3.79	o)
	00	Ampho	ô	$\frac{A1}{2}$	& Ampho	2.21	2.17	1.66	2.84	2.81	2.75	1.79	2.54	1.84	1.83	2.15	1.56	1.80	oteric	3.5	3.58	3.77	erizabl
		Acidic	<u>Oxides</u>	$\frac{\text{sio}}{2}$	1 1/2% to 3	60.2	61.4	61.4	61.8	62.0	62.1	62.7	63.0	63.9	64.1	65.1	9:59	66.7	3 to 4% Amphoteric Oxides	49.8	50.3	55.1	Not Fiberizable
				NO.	1 1/	95	96	97	86	66	100	101	102	103	104	105	106	107	3 to	108	109	110	ii *

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	E-119 Fire West	Thickness	Density		2, 0/2, 12	2.0/1.99	2.0/1.89	2.0/4.02		2.071.93	2 0/1 9	2 0 / 2 0	2.2/2.2	2.0/1.94			1	2.0/1.88	2.0/1.99	2.0/2.00	
5 Hour	Saline	Extraction	pom. Si			ı	19	40	51	9	20	38	2.8	18			37		4	32	;
		Total	Analytical		101.16	100.98	100.09	100.11	101.02	99.41	99.72	. 66.19	69.67	99.38			99.91	100.47	99.91	99.45	ויים
		des	Total		41.85	40.78	39.8	40.28	40.45	0.75 38.55	38.5	0.67 37.17	0.24 37.04	34.34			46.1	39.4	37.55	37.7	= Pass F = Failed
		Basic Oxides	MgO	-	4.65	36.51 4.17	16.2	16.6	4.00	0.75	12.8	0.67	0.24	0.24		-	19.6	9.5	5.65	15.6	. Pass.
WT.8		Bas	<u>Ca0</u>	(Cont.)	37.1	36.51	23.5	23.4	36.45	37.7	25.6	36.4	36.7	34.0			26.4	30.1	31.8	22.0	# b
COMPOSITION, WT&	teric	Oxides	Total	3% to 4% Amphoteric Oxides	3.66	3.65	3.54	3.08	3.64	3.31	3.07	3.77	3.78	3.79		oxides	4.06	5.22	5.41	4.70	
COM	Amphoteric	OX	$\frac{A1}{2}$	photerio	0.24	0.35	3.52	3.06	0.32	3.29	3.05	3.75	3.76	3.77	; ; ;	T CO OF WILDINGLETIC UXIDES	4.04	5.20	5.40	4.68	Not Fiberizable
	Acidic	Oxides	No. $\sin \theta_2$	0 48 Am	55.6	56.5	26.7	26.7	56.88	57.5	58.1	58.2	58.80	61.2	9 7 7	dilla so	49.7	55.8	56.85	57.0	ot Fibe
•	•	٠,	NO.	38 t.	111	112	113	114	1:15	115a	116	117	119	120	4		121	122	123	124	Z *

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		Thickness 1150 Test	Doneitu	Vella LY	ı	2 071 97	2.0/2.0	2.0/3.17	2.0/1.98	2.0/2.04		2.072.01	2.0/2.04		1		1	1	2.071.99		2.0/2.05	_
	Saline	Extraction	is mud		37	. 9	19	18	7	4	2	2	. 0		12	13	_, m	1.2	1.0	1.7	1.2	
		Total	Analytical		98.72	99.83	99.57	99.43	79.69	100.11	100.27	99.93	6.66		100.17	98.69	99.45	101.02	100.05	101.37	100.37	Failed
		des	Total		52.6	45.2	43.8	41.5	37.3	37.6	35.6	35.2	33.1		52.2	46.76	46.12	40.0	37.81	38.9	34.5	F = Fai
		Basic Oxides	MgO		14.0	0.3	18.4	15.2	6.5	6.9	29.7	4.0	5.1		13.7	9.6	0.52	16.2	4.21	16.3	10.9	= Pass,
WT&		Bas	Ca0		38.5	44.8	25.3	26.2	30.7	30.6	5.9	31.2	27.9		38.4	36.7	45.5	23.7	33.5	22.5	23.5	# # p
COMPOSITION,	Amphoteric	Oxides	Total	6 to 8% Amphoteric Oxides	6.92	7.68	6.42	7.48	7.62	6.36	6.72	6.18	7.10	8 to 10% Amphoteric Oxides	9.32	9.13	8.78	8.92	69.6	8.72	9.22	a
CO	Ampho	ô	A1203	hoteric	6.90	7.66	6.40	7.45	7.60	6.34	6.7	6.16	7.08	photeri	9.3	8.8	8.76	6.8	6.67	8.7	9.5	erizable
	Acidic	<u>Oxides</u>	No. Sio ₂	8 Amp	39.5	46.9	49.3	50.4	54.7	56.1	57.9	58.5	59.7	10% Am	38.6	42.8	44.5	52.1	52.5	53.7	56.6	Not Fiberizable
			NO.	6 to	125	126	127	128	129	130	131	132	133	8 to	134	135	136	137	138	139	140	 -

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EXPERIMENTAL DATA

	COM	COMPOSITION,	MT%				5 Hour		
Acidic	Amphoteric	teric					Saline	E-119 Fire Test.	Test.
Oxides	0×	Oxides	Bas	Basic Oxides	des	Total	Extraction	Thickness	2 Hour
1 SiO ₂ to 12% P	Al ₂ 03	. SiO ₂ Al ₂ O ₃ Total to 12% Amphoteric Oxides		MgO	Total	<u>Analytical</u>	ppm, Si	Density	Testar
1 41.0	10.05	10.01	48.25	0.3	48.70	99.87	. 9	2.0/2.00	<u> 24.</u>
2 51.3	10.9	10.92	37.2	0.2	37.5	77.66	0.8	2.0/2.04	14 0.
3 52.4	10.7	10.72	23.1	16.1	39.3	102.42	0.7	2.0/2.00	, : ±.
4 52.7	10.2	10.22	22.1	16.0	38.2	101.12	0.5	I	. l.
to 20% /	\mphoter.	to 20% Amphoteric Oxides				•			
5 41.5	13.0	13.02	44.2	0.5	44.8	99.37	1.2	I	
6 49.8	18.0	18.02	31.5	0.2	32.02	99.89	0.5	ı	J.,
7 55.6	12.9	12.92	13.2	18.4	31.7	100.27	1.8	2.0/2.54	
		٠							
to 30\$	Amphoter	to 30% Amphoteric Oxides					•		~
8 36.5	28.4	28.42	34.4	0.3	34.8	77.66	9.0	i	. J
9 40.3	21.5	21.52	37.5	0.3	37.9	77.66	8.0	ı	_1
0 42.6	25.7	25.72	31.2	0.3	31.6	76.66	9.0	I	. 1
1 48.4	22.4	22.42	16.5	12.6	29.5	100.07	0.5	2.0/2.01	_ 64 _
2 59.9	22.8	22.82	3.1	14.0	17.2	76.66	0.7	2.0/2.01	54 ,
4	1 4 0 1								
TO 408	Amphorer	to 40% Amphoteric Oxides						-	
3 45.9	31.3	31.32	5.9	16.7	22.7	99.97	2.3	1 .	1
= Not Fil	Not Fiberizable	a	:: :: :: :: ::	= Pass,	Ŀ	= Failed			

TABLE 5
FIBERS MADE WITH VARIOUS ADDITIVE CONSTITUENTS

			ANALYSES				5 Hour		
NO.	Acidic NO. Oxides	Amphoteric Oxides	Basic Oxides	Misc.		<pre>% Additive Total (Incl.Total)</pre>	Saline Extraction PPM. Si	Thickness 2 Hour Density Test	re Test. 2 Hour Test.
Fib	ers with	Fibers with B ₂ O ₃ Additions							
164	65.12	90.0	35,3	1	100.48	0.32% B O			
591	64.42	1.20	34.8	1	100.42	0.52% "203	r (2.0/1.94	<u>ට</u> . ෝ
991	65.24	90.0	35.2	1	100.5	0.64%	20	2.0/1.88	2
191	65.32	0.06	35.2	1	100.58	0.82	ል ል ህ የ	2.0/1.89	ച :
.68	65.43	90.0	34.9	ı	100.39	1.338 "		2.0/2.00	<u>.</u>
69.	65.47	90.0	34.9	ı	100.43	1 278	/ **	2.0/1.95	c.
.70	65.82	0.06	y Pt			: 4/5.1	4 2	2.0/ -	a
71	נט פא		•	I	100.48	2.22% "	46	2.0/2.02	d
•		0.0	32.0	ı	100.07	8.418 "	52	2.0/6.45	а
ibe	rs with	ibers with P,Og addition							
72	55.65	0.48	43.58	0.02	99.7	6.06% P205	7.1	2.0/1.94	ţ.
<u>ibe</u> 73	ibers with ?	ibers with TiO ₂ addition							
I) , !		ı	ı	100.	10\$ Tio ₂	0.4	2.01/1.94	d.

			ANALYSES					5 Hour		
								Saline	E-119 Fire Test	re Test
	Acidic	Amphoteric	Basic			% Additive	tive	Extraction	Thickness	2 Hour
NO.	<u>Oxides</u>	Oxides	Oxides	Misc.	Total	(Incl.Total)	otall	ppm. Si	Density_	Test
Fibe	Fibers with 2r0 ₂ -	ZrO ₂ additions	101							
174	63.5	1.10	35.92	1	100.52	0.21%	$2r0_2$	25	2.0/2.01	- 4
175	59.2	0.73	39.51	i	99.44	0.40%	=	48	2.0/2.00	
176	59.5	0.73	39.52	1	99.75	0.42%	=	55	1	ł
177	59.7	0.84	39.16	1	99.70	0.50%	=	32	. 1	1
178	0.09	06.0	38.78	t	89.68	0.54%	=	40	1	ı
179	59.2	0.93	37:98	ı	98.11	0.58%	=	46	2.0/2.02	ď
180	54.3	1.88	43.12	.01	99.31	0.58%	=	29	2.0/2.00	î.
181	59.2	1.15	37.73	ŧ	98.08	0.83%	=	57	2.0/2.03	ď
182	46.85	2.89	49.98	.02	99.74	0.84%	=	44	2.0/2.17	Œ
182a	59.4	2.69	36.96	.02	99.05	2.31%	=	25	2.0/2.00	-
183	59.05	2.95	38.07	·	100.09	2.65%	=	38	2.0/2.20	2
184	57.96	3.53	38.72	1	100.21	3.11%	=	25	2.0/2.37	=
185	57.80	3.68	38.14	ı	99.65	3.12%	=	10	2.0/2.03	í.
981	59.05	3.65	39.51	1	102.21	3.27%	=	15	2.1/2.11	<u>a</u>
187	56.88	3.62	40.45	1	100.95	3.30%	=	51	1	i
188	57.7	3.50	39.0	ı	100.20	3.30%	=	13	2.0/2.06	<u>-</u>
189	58.19	3.75	38.65	1	100.59	3.36%	=	12	. 1	ı
061	57.86	3.73	38.88	1	100.47	3.378	=	ı	2.0/2.00	<u>:-</u>
161	58.6	4.25	36.22	i	99.07	3.67%	=	7	2.0/2.00	<u>-</u>
192	58.4	4.34	. 35.79	I	98.53	3.69%	=	e	2.0/2.00	<u>-</u>
193	58.65	7.87	35.36	.01	99.89	4.50%	=	1.3	2.0/2.07	<u>~</u>

							٠	5 Hour	•	
Test	Test Acidic	Amphoteric	Basic			* Add	% Additive	Saline	E-119 Fire Test	Test
No.	<u>Oxides</u>	Oxides	Oxides	Misc.	Total		(Incl.Total)	Extidction Ppm. Si	Thickness:	2 Hour Test
Fibe	rs with	Fibers with FeO ₁ additions	us						·	
194	64.9	0.06	35.38	ı	100.34	0.06\$	FeO. 6 Mp	u		
195	49.8	18.02	31.92	0.07	99.81	0.22		.,	2.01/1.88	-
196	50.4	7.49	42.04	0.07	100.00	0.52%	=	o. o .	-	i
197	64.34	90.0	34.7	ı	99.1	, C	=	18		1
198	63.70	1.20	33.02	ı	68.62		=	51	2.0/1.91	-
199	63.54	1.20	33.46	ı	20.00 98.00	6,000		24	2.0/1.88	÷
200	38.9	6.72	54.40	0.07	100 00	6 27.0	: :	32	2.0/2.00	1
201	64.3	0.06	35.13	6	60.001	908.0	:	17		ı
200) (06.00	i	100.32	996.0	=	45	2.0/1.88	2
202	44.0	0.94	51.92	ı	97.46	1.02%	=	49		i
203	63.3	1.15	34.99	1	99.44	1.61%	=	12	2 071 96	ä
204	63.6	90.0	36.62	. I	100.15	1.92%	=	<u> </u>	2 0/1 91	. 2
205	43.8	15.28	40.94	0.13	100.02	2.94%	:		17:1	.
206	62.3	1.20	36.05	ı	99.55	3.05%	=		- 0 6	1 -
207	63.3	90.0	36.95	ı	100.31	3.45%	=	` α[2.0/1.90	. .
208	43.9	14.32	41.6	i	99.82	3.50%	=) 1	2.0/1.08	i.
209	62.0	0.06	38.31	1	100.37	4.81%	=	3 ° C .	50 60	1 2
210	0.09	2.0	38.0	ı	100.0	8.0%	=	6	2.0/1.98	: :
211	0.09	1	40.0	1		20.0%	=	7.0	2.0/2.00	: , :

										- 4	: 5 -												
	Teer	2 Hour	<u> 1981.</u>	1	. :	Ŀ	: _	Œ.		2	•		d	. :	<u>.</u>	2.	<u> </u>	à	<u>-</u>	<u>۽</u>	. 14	. 1	-
	E-119 Fire Test	Thickness	_ Density_	 	76.1/0.7	2.0/1.97	2.0/1.98	2.0/1.98		2.000	0.1.2/0.2	-	0 0/1 91	10.1/0.7	2.0/1.97	2.0/1.97	2.0/1.90	2.0/1.90	2.0/1.99	5 071.99	27.17.000	2.0/2.2	2.0/1.8/
5 Hour	Saline	Extraction	ppm. Si		97	69	78	7.0			97		u •	გ	57	54	30	51	57	. 6	ਹ † †	20	70
		tive	<u>rotall</u>		La ₂ 0 ₃	=	=	=			cr_{2}^{0} 3		,	Na_2O	=	=	=	=	=	: :	=	=	=
Ì		% Additive	(Incl.Total)		0.00 % La $_2$ 0 $_3$	0.56%	0.72%	900	9.76		0.09% Cr ₂ 03		,	0.28%	0.45%	0.71%	0.87%	0.93%	0 0	1.116	1.40%	2.60%	6.84%
			Misc. Total		99.63	89.66	99.28	¥ 4 0 0	40.66		99.72			100.34	100.21	100.26	100.40	66 66		100.37	100.36	100.06	1001
S					ı	ı	ı		ı		i			1	ı	1	ı	1	l	ı	ı	i	1
ANALYSES		Basic	•	Suo	41.47	41.82	41 72	1 · · ·	41.58	ions	36.61		ons	35.58	35.68	35.80	25.25		15.61	36.11	36.3	37.0	39.74
		Amphoteric	Oxides	ibers with La,0, additions	90.0	0.06	90.0	0.00	90.0	ibers with Cr ₂ 0 ₃ additions	0.51	•	ibers with Na ₂ O additions	0.06	90 0	90.0	0 6	1.20	90.0	90.0	90.0	0.06	90.0
		טינטיטע דיים	Oxides	with	58.1	57 B		o./c	56.9	s with	62.6		s with	64.7			# · · · ·	63.5	64.3	64.2	64.0	63.0	60.3
	1	+ U		iber	12		7 .	14	15	iber	16		iber	713		0 0	61;	520	121	322	223	224	225

	St.		-46-	
	re Test 2 Hour Test.		rraaa111111	÷
	E-119 Fire Test Thickness 2 Hour Density Test	2.0/3.50 2.0/5.23 2.0/3.42 2.0/3.86	2.0/2.10 2.0/5.38 2.0/2.00 2.0/2.00 	2.0/1.85
5 Hour	Saline Extraction ppm. Si	7 1.2 0.6 1.0	2 0.6 0.8 0.3 1.0 0.4 0.3 0.4	0.8
	<pre>% Additive (Incl.Total)</pre>		less than 25% Basic Oxides) - 99.92 99.6 100 100 99.62 99.62 99.65 99.65 99.78 100.23 100.23 -	ı
	<u>Total</u>	100.16 100.47 100.69 101.14	an 25% Ba 99.92 99.6 100 100.11 99.62 99.91 100.04 99.65	100.3
SES	Misc.	0.69 0.74 0.61 0.64	less that the state of the stat	ı
ANALYSES	Basic Oxides 1 Fibers	49.97 45.82 49.35 41.53	with 4 4 5 7 7 7 8 8 8	4.0
	Test Acidic Amphoteric Basic No. Oxides Oxides Conventional Mineral Wool Fibers	9.50 13.99 12.24 17.10		F
	Test Acidic No. Oxides Conventional	40.0 39.92 38.49 41.87	Refractory Fibers - 231 31.0 47.52 232 37.1 59.2 233 50.0 40.0 334 54.0 46.0 335 59.62 25.55 36 52.1 46.84 37 52.0 46.84 39 48.6 50.05 40 47.8 51.00 41 46.2 53.10 42 28 72 43 64.5 27.4	•
	Test No. Conve	226 227 228 228	Refra 231 232 233 234 234 336 340 41 42	

TABLE 6

CONTINUOUS SERVICE TEMPERATURE FOR CONSTANT SiO₂/CaO/MgO RATIOS

30	Continuous Service Temperature for max 5% shrinkage		1550	1520	1480	1600	1520	1500
20	for max 5		1420	1400	1350	1460	1410	1350
10	<u>remperature</u>	ů.	1470	1420	1370	1460	1400	1360
- 22	Service		1480	1430	1380	1460	1420	1370
0	Continuous		1480	1440	1400	1500	1430	1380
	Sio,/CaO/MqO Ratio	7	50/50/0	50/40/10	50/10/10	60/40/0	60/30/10	60/20/20

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Reasonable modifications and variations are possible from the foregoing disclosure without departing from either the spirit or scope of the invention asdefined in the claims.

CLAIMS

	1. A process for decomposing a silica-
	containing fiber comprising the steps of:
	 providing an inorganic fiber pre-
5	pared from a composition consisting essential-
	ly of:
	(a) 0.06-10 wt% of a material
	selected from the group consisting of
	Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and
10	mixtures thereof;
	(b) 35-70 wt% SiO ₂ ;
	(c) 0-50 wt% MgO; and
	(d) the remainder consisting essen-
	tially of CaO, the total being 100% by
15	weight;
	2. subjecting the silica-containing
	fiber to a physiological caling
	fiber to a physiological saline fluid; and
	 extracting the silica at a rate of
	at least 5 parts per million (ppm) of silicon
20	in 5 hours, thereby decomposing the silica-
	containing fiber.

- The process of Claim 1 wherein the composition of subsection 1(a) ranges from 0.06-5 wt% of material selected from the group consisting of Al₂O₃,
 ZrO₂, TiO₂, B₂O₃, iron oxides and mixtures thereof.
 - 3. The process of Claim 1 wherein the composition of subsection l(c) ranges from 0.25-50 wt% MgO.
- 4. The process of Claim 1 wherein the composition consists essentially of:

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	(a)	0.06-1.	5 Wt%	of	A1.0) ₃ , 2ro ₃ ,
TiO_2 ,	B ₂ O ₃ ,	iron	oxides	a	nd -	mixtures
there	of;					342 43

- (b) 40-70 wt% SiO₂;
- (c) 0-50 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100% by weight.
- 5. The process of Claim 4 wherein the composition in subsection 1(c) ranges from 0.25-50 wt% MgO.
 - 6. The process of Claim 1 wherein the composition consists essentially of:
 - (a) 1.5-3 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 40-66 wt% SiO2;
 - (c) 0-50 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100% by weight.
 - 7. The process of Claim 1 wherein the composition of subsection 1(c) ranges from 0.25-50 wt% MgO.
- 8. The process of Claim 1 wherein the composition consists essentially of:
 - (a) 3-4 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 40-63 wt% SiO,;
 - (c) 0-50 wt% MgO; and

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(c	i)	the re	main	der cor	nsistin	g ess	≘n-
tially	of	CaO,	the	total	being	100%	уď
weight.	•						

- 9. The process of Claim 8 wherein the composition of subsection 1(c) ranges from 0.25-50 wt% MgO.
 - 10. The process of Claim 1 wherein the composition consists essentially of:
 - (a) 4-6 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 40-60 wt% SiO₂;
 - (c) 0-25 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100% by weight.
 - 11. The process of Claim 10 wherein the composition of subsection 1(c) ranges from 0.25-25 wt% MgO.
- 12. The process of Claim 1 wherein the 20 composition consists essentially of:
 - (a) 6-8 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 35-54 wt% Sio,;
 - (c) 0-25 wt% MgO; and
- 25 (d) the remainder consisting essentially of CaO, the total being 100% by weight.
 - 13. The process of Claim 12 wherein the composition of subsection 1(c) ranges from 0.25-25 wt% MgO.

- 14. The process of Claim 1 wherein the composition consists essentially of:
 - (a) 3-10 wt% of Al_2O_2 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
- (b) 35-54 wt% SiO₂;
 - (c) 0-20 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100% by weight.
- 15. The process of Claim 14 wherein the composition of subsection 1(c) ranges from 0.25-20 wt% MgO.
 - 16. The process of Claim 1 wherein the fiber has a diameter of less than 3.5 microns.
- 17. The process of Claim 1 wherein the silicon extraction rate is at least 20 ppm, the Al₂O₃ content is about 0.06-7 wt%, and the SiO₂ content is about 40-66 wt%.
- 18. The process of Claim 1 wherein the silicon extraction rate is at least about 50 ppm, the Al_2O_3 content is about 0.06-3 wt%, and the SiO_2 content is about 40-60 wt%.
- 19. The process of Claim 1 wherein the silicon extraction rate is at least about 50 ppm, the Al₂O₃ content is about 0.06-0.75 wt%, and the SiO₂ content is about 40-60 wt%.
 - 20. A process of protecting a structural wall from fire comprising the steps of:

	 providing a fiber blanket having a
	bulk density in the range of about 1.5 to
	about 3 lbs. per cubic foot (pcf); wherein the
	fiber blanket has the ability-to pass ASTM
5	E-119 two-hour fire test; the fibers in the
3	blanket have a diameter less than about 3.5
	microns; and the fiber is an inorganic fiber
	prepared from a composition consisting essen-
	tially of:
10	(a) 0-7 wt% of Al_2O_3 , ZrO_2 , TiO_2 ,
10	B_2O_3 , iron oxides and mixtures thereof;
	(b) 58-70 wt% SiO₂
	(c) 0-21 wt% MgO;
	(d) 0-2 wt% alkali metal oxide; and
15	(e) the remainder consisting essen-
10	tially of CaO, the total being 100% by
	weight; and
	 placing the blanket next to the
	wall, and thereby protecting the wall from
20	fire.

- 21. The process of Claim 20 wherein the composition of subsection 1(a) ranges from 0.06-7 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof.
- 22. The process of Claim 20 wherein the composition of subsection 1(c) ranges from 0.25-21 wt% MgO.
 - 23. The process of Claim 20 wherein the composition consists essentially of:
 - (a) 0.06-3.0 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 58.5-70 wt% SiO₂;

SUBSTITUTE SHEFT

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(C)	0-21	wtg	MgO;
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- (d) 0-2 wt% alkali metal oxide; and
- (e) the remainder consisting essentially of Cao, the total being 100% by weight.
- 24. The process of Claim 20 wherein the composition of subsection 1(c) ranges from 0.25-21 wt% MgO.
- 25. The process of Claim 20 wherein the composition consists essentially of:
 - (a) from about 3 wt% up to and including 4 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 58-63 wt% SiO₂;

0-8 wt% MgO;

- (c)
 - (d) 0-2 wt% alkali metal oxide; and
 - (e) the remainder consisting essentially of CaO, the total being 100% by weight.
- 26. The process of Claim 25 wherein the composition in subsection 1(c) ranges from 0.25-8 wt% MgO.
 - 27. The process of Claim 25 wherein the composition consists essentially of:
- (a) from about 4 wt% up to and including 6 wt% of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 58-61 wt% SiO₂;
 - (c) 0-7 wt% MgO;
 - (d) 0-2 wt% alkali metal oxide; and

- (e) the remainder consisting essentially of Cao, the total being 100% by weight.
- The process of Claim 25 wherein the composition of subsection 1(c) ranges from 0.25-7 wt% 5 MgO.
 - An inorganic fiber having an average fiber diameter of less than about 3.5 microns, a silicon extraction rate greater than about 0.02 wt% Si/day in a physiological saline solution and having a composition consisting essentially of about:
 - 0.06-5.0 wt% of (a) selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 35-70 wt% SiO₂;
 - (c) 0-50 wt% MgO; and
 - (d) the remainder consisting essentially of Cao, the total being 100 wt%.
- 30. An inorganic fiber having a silicon 20 extraction of at least about 10 ppm over a 5 hour period in physiological saline solution and having a composition consisting essentially of about:
 - 0.06-1.5 wt% of material (a) selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 40-70 wt% SiO₂;
 - (c) 0-50 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100 wt%.

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- 31. An inorganic fiber according to Claim 30 having a silicon extraction of at least about 20 ppm, an average fiber diameter of less than about 3.5 microns, and having an SiO₂ content of about 40-66 wt%.
- 5 32. An inorganic fiber according to Claim 30 having a silicon extraction of at least about 50 ppm and having an SiO₂ content of about 40-60 wt% and a MgO content of about 0.25-25 wt%.
- 33. An inorganic fiber having a silicon extraction of at least about 10 ppm over a 5 hour period in physiological saline solutions and having a composition consisting essentially of about:
 - (a) 1.5-3 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof:
 - (b) 40-66 wt% Sio,;
 - (c) 0-50 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100 wt%.
 - 34. An inorganic fiber according to Claim 33 having a silicon extraction of at least about 20 ppm, an average fiber diameter of less than about 3.5 microns, and an MgO content of from about .25-50 wt%.
- 25 35. An inorganic fiber according to Claim 33 having a silicon extraction of at least about 50 ppm, an SiO₂ content of from about 40-54 wt%, and an MgO content of from about 0.25-18 wt%.
- 36. An inorganic fiber having a silicon30 extraction of at least about 10 ppm over a 5 hour period

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in physiological saline solutions and having a composition consisting essentially of about:

- (a) 3-4 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 40-63 wt% SiO₂;
 - (c) 0-50 wt% MgO; and
- (d) the remainder consisting essentially of CaO, the total being 100 wt%.
- 37. An inorganic fiber according to Claim 36 having a silicon extraction of at least about 20 ppm, an average fiber diameter of less than about 3.5 microns, and a SiO₂ content from about 40-58 wt%.
- 15 38. An inorganic fiber according to Claim 37 having a silicon extraction of at least about 50 ppm and an SiO₂ content of from about 40-52 wt% and a MgO content of from about .25-18 wt%.
- 39. An inorganic fiber having a silicon extraction of at least about 10 ppm over a 5 hour time period in a physiological saline solution and having a composition consisting essentially of about:
 - (a) 4-6 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof:
 - (b) 40-59 wt% SiO₂;
 - (c) 0-46 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100 wt%.

40. An inorganic fiber according to Claim	30
having a silicon extraction of at least about 20 ppm,	an
average fiber diameter of less than about 3.5 micron	s.,.
and an SiO ₂ content from about 40-58 wt%.	

- 41. An inorganic fiber having a diameter of 5 less than about 3.5 microns and which passes the ASTM E-119 two hour fire test when processed into a fiber blanket having a bulk density in the range of about 1.5 to 3 pcf, said inorganic fiber having a composition consisting essentially of:
 - (a) .06-7 wt% of material selected from the group consisting of Al₂O₃, ZrO₂, TiO₂, B₂O₃, iron oxides and mixtures thereof:

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- (b) 58-70 wt% SiO,;
- (c) 0-21 wt% MgO;
- 0.1-2 wt% alkali metal oxide; (d)

and

(e) the remainder consisting essentially of CaO, the total being 100 wt%; wherein the amount of alumina + zirconia is less than 6 wt% and the amount of iron oxides or alumina + iron oxides is less than 2 wt%.

- An inorganic fiber according to Claim 41 having a composition consisting essentially of about: 25
 - .06-1.5 wt% of material se-(a) lected from the group consisting of Al₂O₁, ZrO₂, TiO₂, B₂O₃, iron oxides and mixtures thereof; and

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(b) 58.5-70 wt% SiO₂.

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- 43. An inorganic fiber according to Claim 42 having a silicon extraction of at least about 10 ppm over a 5 hour period in physiological saline solutions.
- 44. An inorganic fiber according to Claim 41 having a composition consisting essentially of about:
 - (a) greater than 1.5 wt% up to and including 3 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof; and
 - (b) 58.5-66 wt% SiO₂.
 - 45. An inorganic fiber according to Claim 44 having a silicon extraction of at least about 10 ppm over a 5 hour period in a physiological saline solution.
- 15 46. An inorganic fiber according to Claim 41 having a composition consisting essentially of about:
 - (a) greater than 3 wt% up to and including 4 wt% material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 58-63 wt% SiO₂;
 - (c) .25-8 wt% MgO;
 - (d) .1-2 wt% alkali metal oxide;

and

- (e) the remainder consisting essentially of CaO, the total being 100 wt%.
- 47. An inorganic fiber according to Claim 46 having a silicon extraction of at least about 10 ppm over a 5 hour period in physiological saline soluti ns.



		48. 2	'n.	inc	organic	fiber	according	to	Claim	4 1
having	a	composi	iti	.on	consist	ing es	ssentially	οf	about	

- (a) greater than 4 wt% up to and including 6 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 58-59 wt% SiO₂;
 - (c) .25-7 wt% MgO;
 - (d) .1-2 wt% alkali metal oxide;

10 and

- (e) the remainder consisting essentially of CaO, the total being 100 wt%.
- 49. An inorganic fiber according to Claim 48 having a silicon extraction of at least about 10 ppm over a 5 hour period in physiological saline solutions.
 - 50. An inorganic fiber having a silicon extraction of greater than about 0.02 wt% Si/day in a physiological saline solution, a continuous service temperature above about 1450°F and having a composition consisting essentially of about:
 - (a) .06-5 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof:

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- (b) 40-70 wt% SiO₂;
- (c) 0-6 wt% MgO; and
- (d) the remainder comprising essentially of CaO, the total being 100 wt%.
- 51. The fiber of Claim 50 wherein the composition of subsection (c) has an amount of 0.25-6 wt% MgO.

- 52. An inorganic fiber having a silicon extraction of greater than about 0.02 wt% Si/day in a physiological saline solution, having a continuous service temperature above about 1500°F and having a composition consisting essentially of about:
 - (a) .06-1.5 wt% of material selected from the group consisting of Al_2O_3 , ZrO_2 , TiO_2 , B_2O_3 , iron oxides and mixtures thereof;
 - (b) 60-70 wt% SiO₂;
 - (c) 0-1 wt% MgO; and
 - (d) the remainder consisting essentially of CaO, the total being 100 wt%.
- 53. The fiber of Claim 52 wherein the compo-15 sition of subsection (c) has an amount 0.25-1 wt% MgO.
 - 54. An inorganic fiber according to Claims 1 or 29 made from pure oxidic raw materials.
- or 29 or 41 in which at least a portion of the raw materials is selected from a group consisting of talc, metallurgical slags, siliceous rocks, kaolin, and mixtures thereof.
 - 56. An inorganic fiber having a composition consisting essentially of about:
 - (a) 8.0-9.3 wt% Al₂O₃;
 - (b) 39-52 wt% SiO₂;
 - (c) 22-38 wt% CaO; and
 - (d) 7-14 wt% MgO, the total being 100 wt% and having a silica extraction in a saline solution of at least about 5 ppm over a 5 hour period.

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